

Synthesis of Thiazoles and 1,3,4-Thiadiazoles Bearing Spectral Studies, Biological Evaluation and Structure Activity relationship

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Abstract— A novel series of thiazole based-1,3,4-thiadiazoles were designed and prepared via the reaction of the 2-(4-methyl-2-phenylthiazole-5-carbonyl)-N-phenylhydrazinecarbothioamide with the appropriate hydrazonoyl chlorides. The structures of the newly synthesized compounds were established based on spectroscopic evidences and their alternative syntheses. Thirteen new 1,3,4-thiadiazoles have been evaluated for their anticancer activity against liver carcinoma cell line (HepG2). Also, their structure activity relationship (SAR) was studied. The 1,3,4-thiadiazoles **12d**, **12c**, **6g**, **18b**, **6c**, and **6f** (IC_{50} = 0.82, 0.91, 1.06, 1.25, 1.29 and 1.88 μ M, respectively) have promising antitumor activity against liver carcinoma cell line (HepG2).

Keywords— thiazoles; thiadiazoles; hydrazonoyl chlorides; anticancer activity; structure activity relationship.

I. INTRODUCTION

Cancer is a devastating and most common life-threatening disease representing a major health problem for many decades. The clinical application of chemotherapy still considered as a major compartment in treating cancer, however, is often limited by the severity of the side effects and the development of tumor cell resistance against these cytotoxic agents. Clinical administration of high doses of anticancer drugs to overcome resistance leads to severe toxicities [1]. Therefore, the development of novel effective anticancer drugs and strategies is eagerly being pursued.

Also, it was reported that Liver cancer is ranked in the top ten human cancers worldwide and among the top five of cancers in terms of mortality [2,3]. A literature survey revealed that a great deal of interest has been focused on the synthesis of functionalized thiazole derivatives due to their synthetic and biological potentialities as antihypertension [4], antifungal [5], antimicrobial [6,7], anti-inflammatory [8], antioxidant [9], antitubercular [10], and anticancer agents [11-14]. 1,3,4-Thiadiazole derivatives have attracted considerable interest due to their wide spectra of biological activities such as antibacterial, antifungal, antituberculosis, antihepatitis B viral, antileishmanial, anti-inflammatory, analgesic, CNS depressant, antioxidant, antidiabetic, molluscicidal, antihypertensive, diuretic, analgesic, antimicrobial, antitubercular, anticonvulsant and anticancer activities [15–24]. These important biological activities encouraged several research groups to find out different methods for synthesis of new thiadiazoles using different syntheses, such as thiosemicarbazides, thiocarbazides, dithiocarbazates, thioacylhydrazines, acylhydrazines, and bithioureas [25]. In the light of the above-mentioned findings and in continuation of our efforts to synthesize new bioactive compounds [26-34], this work aims to synthesis a new series of thiazoles and 1,3,4-thiadiazoles bearing thiazole moiety and to study their anticancer activity against Liver carcinoma cell line (HepG2).

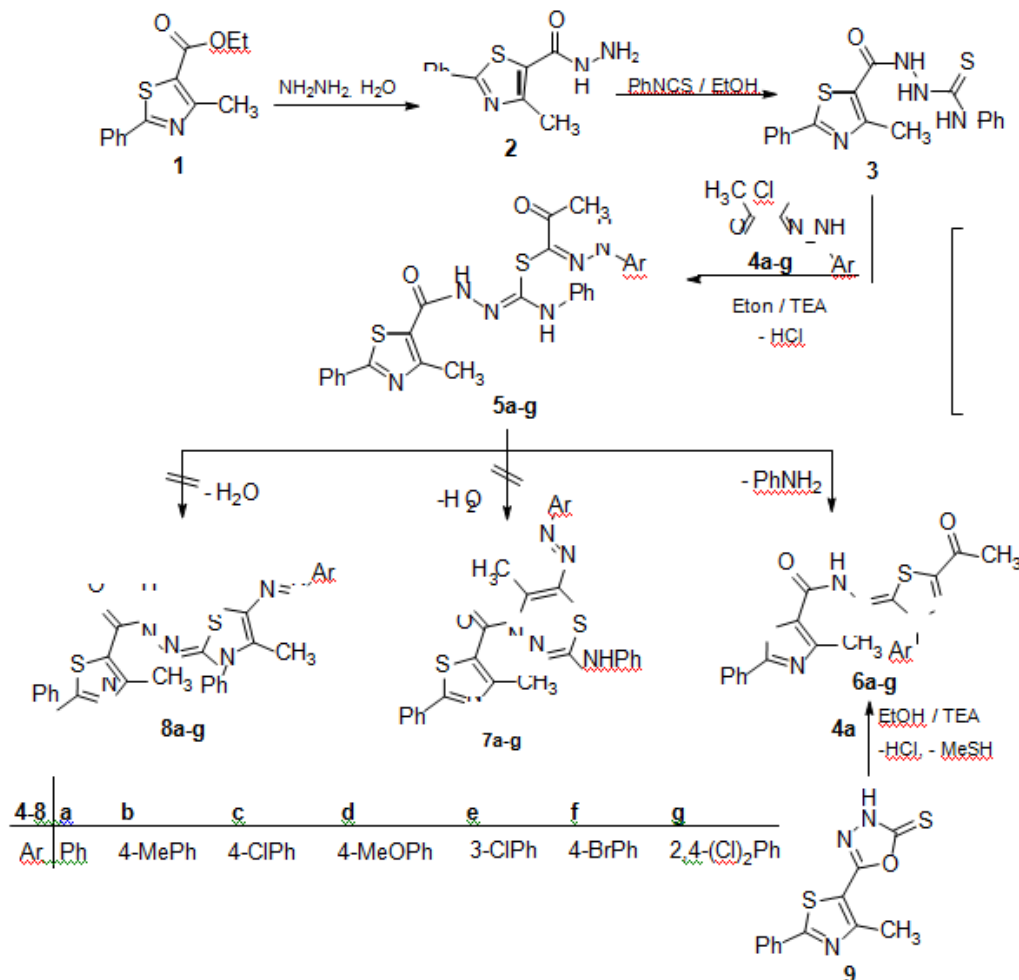
II. RESULTS AND DISCUSSION

2.1 Chemistry

2-(4-Methyl-2-phenylthiazole-5-carbonyl)-N-phenylhydrazinecarbothioamide (**3**) was prepared as previously described, via reaction of 4-methyl-2-phenylthiazole-5-carbohydrazide (**2**) with phenyl isothiocyanate in EtOH as depicted in Scheme 1 [35].

The presence of the thioamidehydrazine moiety as a side chain in compound **3** prompted us to utilize it for constructing 1,3,4-thiadiazole ring *via* its reaction with hydrazonoyl chlorides. Thus, reaction of compound **3** with the appropriate 2-oxo-*N'*-arylpropanehydrazonoyl chlorides **4a-g** [36] under reflux in ethanol in the presence of triethylamine as a basic catalyst led to formation of the respective 1,3,4-thiadiazoles **6a-g**, rather than thiadiazines **7a-g** or 1,3-thiazoles **8a-g** (Scheme 1). The elemental analysis together with the data derived from IR, ¹HNMR and mass spectra are in agreement with the proposed structure **6**. The IR spectra of products **6** showed in each case the presence of two absorption bands around 1700, 1650 cm^{-1} for the two carbonyl groups, in addition to another band near ν 3350 cm^{-1} for the NH function. The ¹HNMR spectra of **6**

revealed the presence of broad singlet signals assigned for the NH proton near δ 11.19 ppm, in addition to the expected signals for the protons of the CH₃ group, the acetyl group at position-2 of the 1,3,4-thiadiazole ring and the aryl protons. The mass spectrum of each of products **6** revealed the presence of a molecular ion peak (see experimental section).



SCHEME 1. Synthesis of thiadiazoles 6a-g

1,3,4-Thiadiazole **6a-g** was assumed to be formed through the intramolecular cyclization of NH group in the hydrazone moiety with the imino group in the non-isolable intermediates **5a-g**, followed by elimination of aniline molecule to give the respective thiadiazole derivatives **6a-g** (Scheme 1).

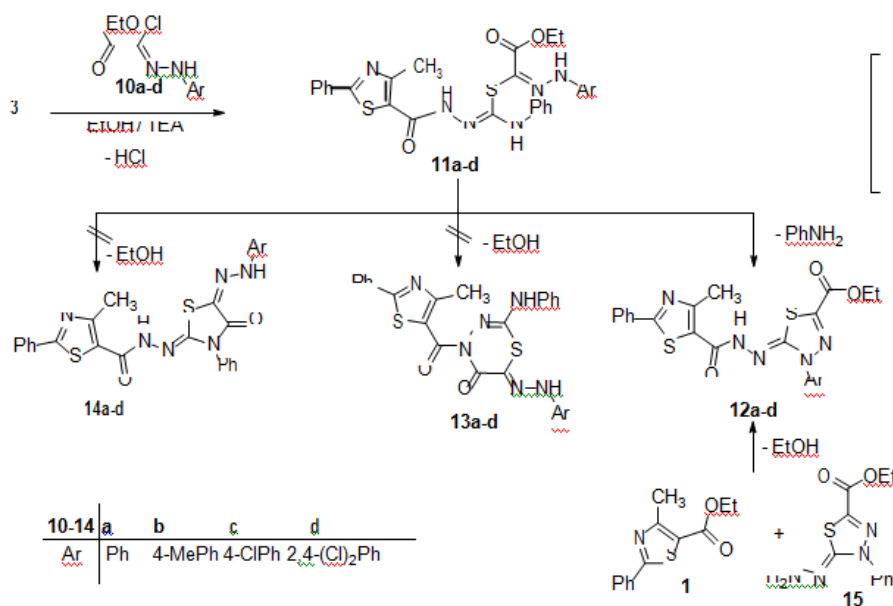
The structure of **6** was proved chemically *via* an alternative method (Scheme 1). Thus, the reaction of 5-(4-methyl-2-phenylthiazol-5-yl)-1,3,4-oxadiazole-2(3*H*)-thione(**9**) [35] with **4a** in ethanol in the presence of triethylamine under reflux led to formation of product which is identical in all respects (mp, mixed mp and IR) with compound **6a**.

Next, in order to test of the biological activities of a vast array of these compounds, we reacted compound **3** with the appropriate hydrazonoyl chlorides **10a-d** [36], under the same experimental conditions, which gave the corresponding 1,3,4-thiadiazole derivatives **12a-d** (Scheme 2).

The IR, and ¹H-NMR spectra of **12a** taken as an example of the prepared series, revealed the presence of the ester group and the disappearance of the hydrazone-NH function. Also, the mass spectrum of the reaction products **12a-d** showed, in each case, a peak corresponding to their molecular ions.

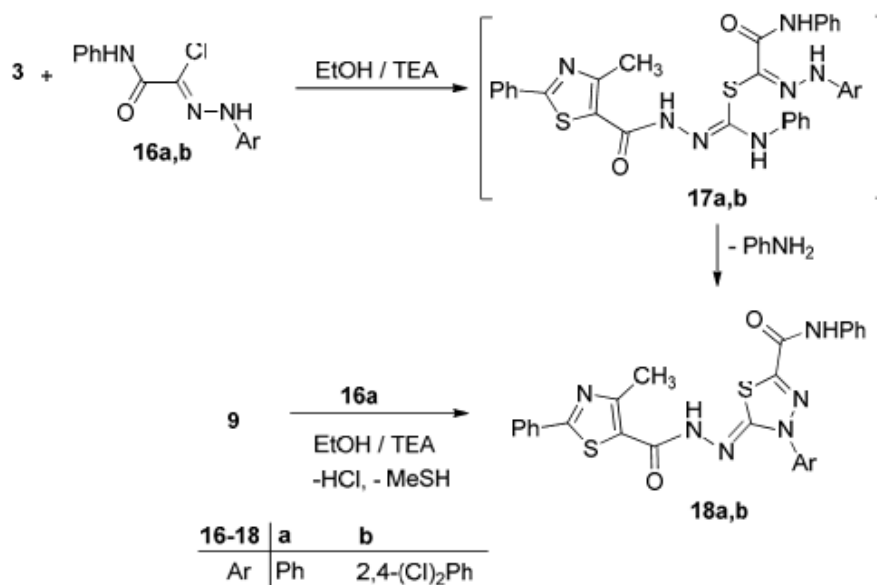
The structure assigned for product **12** was further evidenced *via* an alternative method. Thus, reaction of ethyl 4-methyl-2-phenylthiazole-5-carboxylate (**1**) with 1,3,4-thiadiazole **15** [37] in ethanol under reflux, afforded a product which is typical in all respects (mp, mixed mp and IR) with that obtained from the reaction of **3** with **10a** (Scheme 2). To account for the formation of the product **12**, it was suggested that the reaction of compound **3** with hydrazonoyl chloride **10** initially gave the intermediate **11**, which underwent nucleophilic addition, followed by *in situ* cyclization *via* losing of one molecule of aniline

(route a) to give the final product **12**. The other routes (b) and (c) outlined in Scheme 2 were excluded since they led to formation of products **13** and **14**, which were completely different in all respects (IR, ^1H NMR, mass spectra) from products **12**.



SCHEME 2. Synthesis of thiadiazole derivatives 12a-d

Also, The reaction of compound **3** with hydrazonoyl halide of type **16** was studied. Thus refluxing compound **3** with the hydrazonoyl chloride **16a** or **16b** [36] under the same experimental conditions, afforded the corresponding 1,3,4-thiadiazole derivatives **18a,b** (Scheme 3). The ^1H NMR spectrum of compound **18a**, revealed two D_2O -exchangeable signals at δ 10.18 and 11.72 corresponding to two NH protons, in addition to an aromatic multiple in the region 7.02-7.78 ppm. Also, its mass spectrum of revealed a molecular ion peak at $m/z = 512$ which is in complete agreement with the proposed structure (see Experimental). In addition, compound **18a** was proved chemically *via* an alternative method from the reaction of compound **9** with **16a** which gave a product identical in all respects (mp, mixed mp and IR) with compound **18a**.

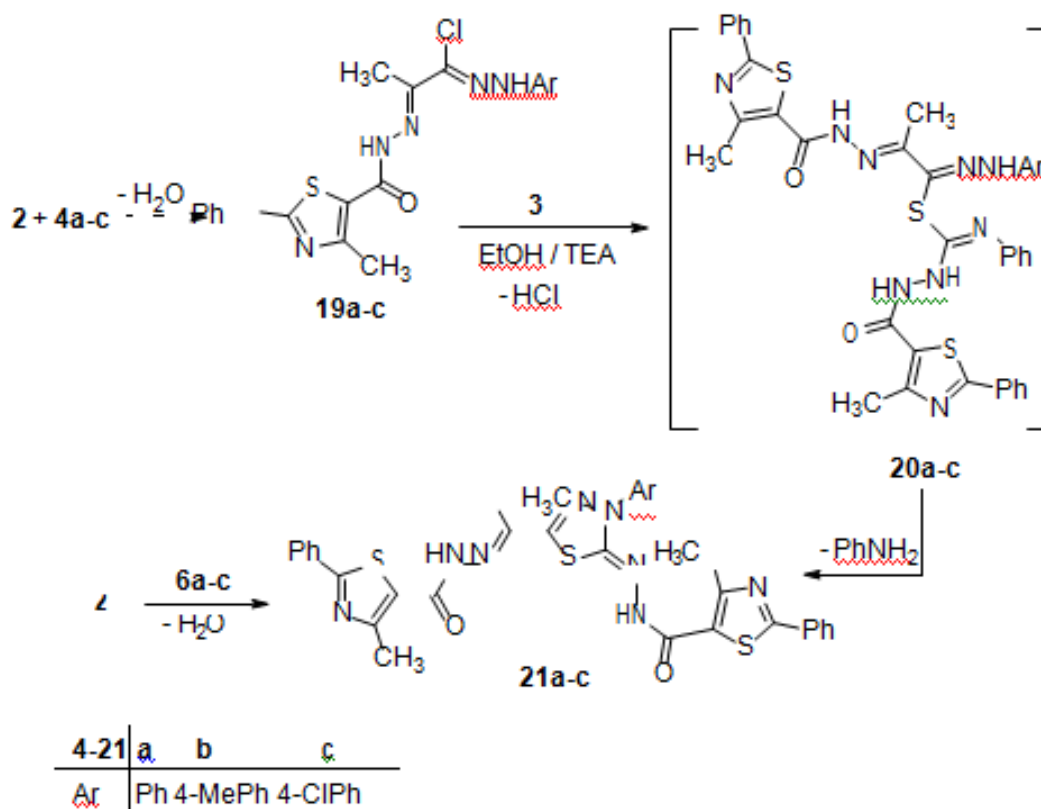


SCHEME 3. Synthesis of thiadiazole derivatives 18a,b

The reaction of 4-methyl-2-phenylthiazole-5-carbohydrazide (**2**) with 2-oxo-*N*-arylpropane hydrazonoyl chlorides **4a-c** in refluxing ethanol gave, in each case, the corresponding condensation product whose elemental analysis and spectra data(see

Experimental) were consistent with structure **19** (Scheme 4). The IR spectra of the latter products exhibited a carbonyl and two NH absorption bands (see experimental part). Their $^1\text{H-NMR}$ showed two D_2O exchangeable signals of two NH groups in the regions δ 10.03–10.06 and δ 10.57–10.59 ppm. Also, the mass spectra of the latter products confirmed the assigned structure **19** (Scheme 4).

Treatment of thioamide derivative **3** with the appropriate hydrazonoyl halides of type **19a-c** in refluxing EtOH in the presence of TEA gave the corresponding thiadiazole derivatives **21a-c** (Scheme 4). The structures of the isolated products **21a-c** were elucidated on the basis of their spectral data and elemental analysis (see Experimental section). The latter products **21a-c** were alternatively prepared by condensing **6a-c** each with 4-methyl-2-phenylthiazole-5-carbohydrazide (**2**) in refluxing ethanol in quantitative yields (Scheme 4).



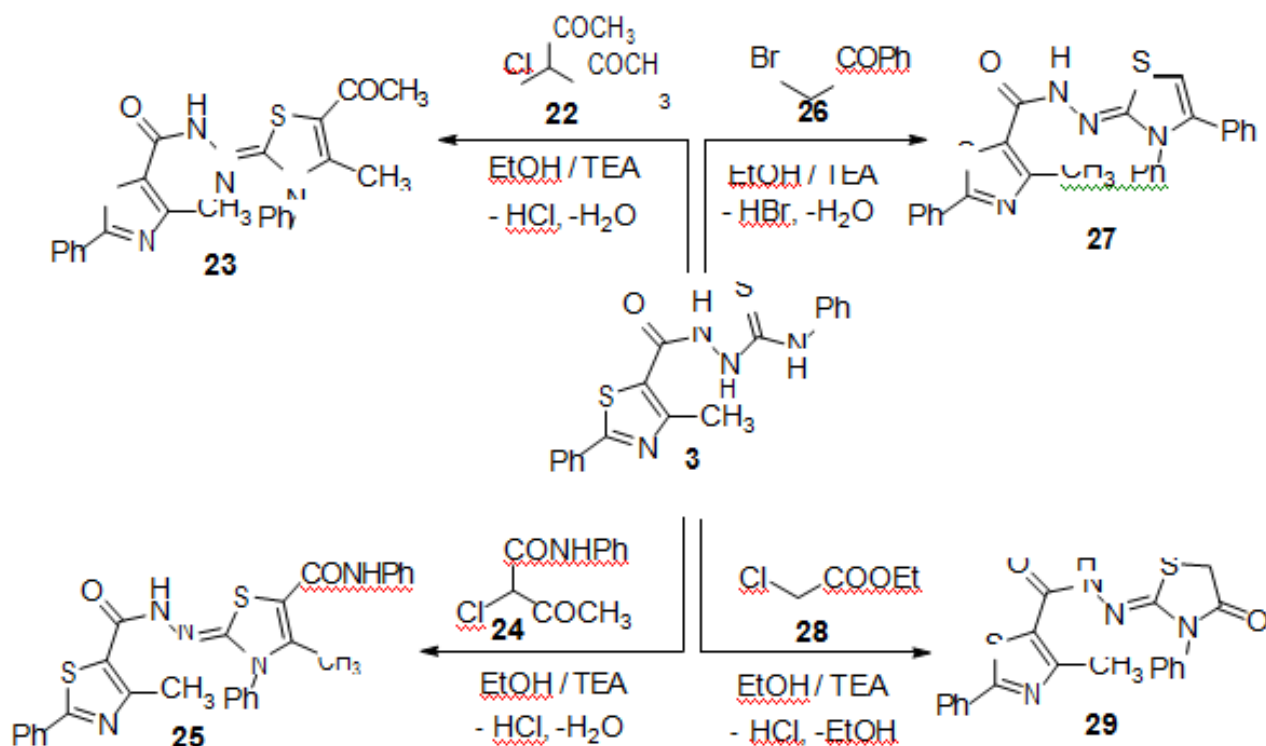
SCHEME 4. Synthesis of thiadiazole derivatives 21a-c

Finally, the starting compound **3** was used for preparation of thiazole derivatives.

Thus, refluxing of compound **3** with 3-chloropentane-2,4-dione (**22**) or 2-chloro-3-oxo-N-phenylbutanamide (**24**) in EtOH in the presence of triethylamine afforded the thiazole derivatives **23** and **25**, respectively, as outlined in Scheme 5. The structure of compounds **23** and **25** were elucidated based on their elemental analysis and spectral data (see Experimental).

Also, thioamide derivative **3** reacted with phenacyl bromide **26** under the same experimental condition to afford one isolable product **27** named as *N'*-(3,4-diphenylthiazol-2(3*H*)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide (Scheme 5). The structure of the product **27** was established based on its elemental analysis and spectral data (see Experimental).

In a similar manner, thioamide derivative **3** reacted with ethyl chloroacetate (**28**) to afford a single product **29** that was identified as 4-methyl-*N'*-(4-oxo-3-phenylthiazolidin-2-ylidene)-2-phenylthiazole-5-carbohydrazide (**29**) as outlined in Scheme 3. The structure of the isolated product **29** was established from its elemental analysis and spectral data. Its IR spectrum showed absorption bands at ν 3331 (NH), and 1726, 1648 ($2\text{C}=\text{O}$) cm^{-1} , its $^1\text{H-NMR}$ spectrum showed singlet signal at δ 4.23 ppm due to the thiazolidinone (CH_2) group (see Experimental section).



SCHEME 5. Synthesis of thiazole derivatives 23, 25, 27 and 29.

2.2 Cytotoxic activity

The Literature survey showed that many derivatives of thiazole and 1,3,4-thiadiazole have antitumor activity with excellent IG₅₀ as depicted in Figure 1 [38-42].

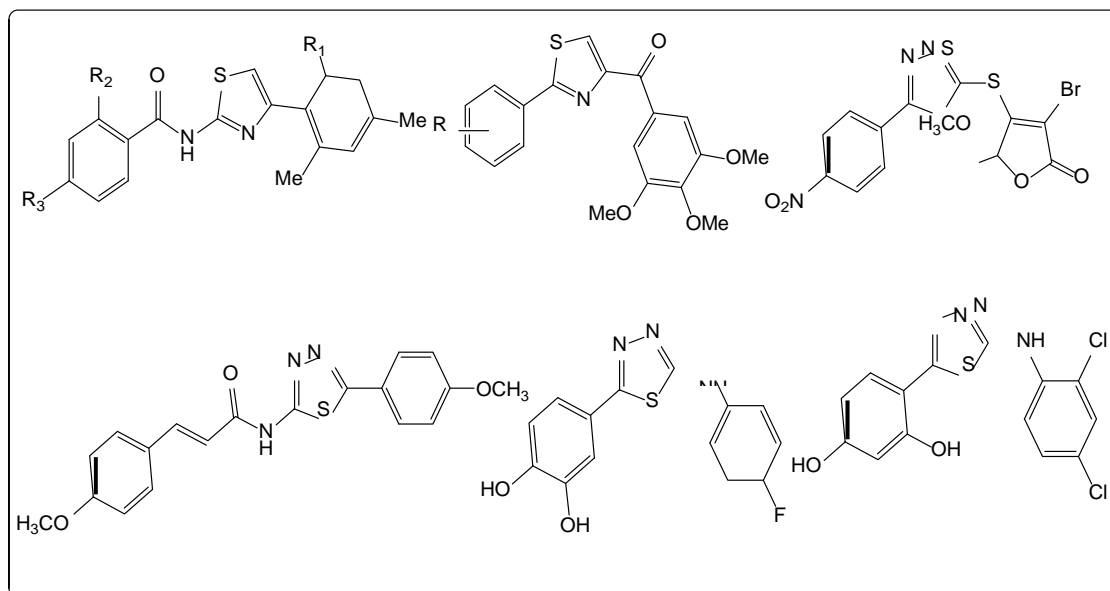


FIGURE 1. Lead compounds among thiazole and thiadiazole derivatives with anticancer activity

In view of these facts, the antitumor activity of the synthesized compounds was determined against a liver carcinoma cell line HepG2. Doxorubicin was used as a reference standard and showed IC₅₀ = 0.72 μM against a liver carcinoma cell line. Data generated were used to plot a dose-response curve of which the concentration (μM) of test compounds required to kill 50% of cell population (IC₅₀) was determined. Cytotoxic activity was expressed as the mean IC₅₀ of three independent experiments. The results depicted in table 1.

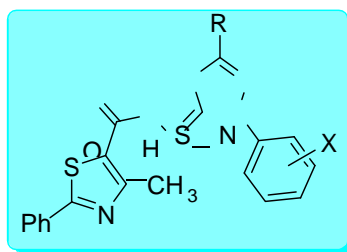
TABLE 1
CYTOTOXIC ACTIVITIES OF TESTED COMPOUNDS AGAINST LIVER CARCINOMA CELL LINE (HEPG2)

Sample Number	R	X	IC ₅₀ (μM)
Doxorubicin	-----	-----	0.72
6a	Ac	H	9.89
6b	Ac	4-Me	39.06
6c	Ac	4-Cl	1.29
6d	Ac	4-OMe	64.35
6e	Ac	3-Cl	4.03
6f	Ac	4-Br	1.88
6g	Ac	2,4-(Cl) ₂	1.06
12a	CO ₂ Et	H	4.70
12b	CO ₂ Et	4-Me	32.46
12c	CO ₂ Et	4-Cl	0.91
12d	CO ₂ Et	2,4-(Cl) ₂	0.82
18a	CONHPh	H	6.79
18b	CONHPh	2,4-(Cl) ₂	1.25

The results revealed that most of the tested compounds showed a great variable activity compared to reference drug as shown in Table 1. The order of activity of the newly synthesized compounds was as follow: **6d < 6b < 12b < 6a < 18a < 12a < 6e < 6f < 6c < 18b < 6g < 12c < 12d**

These results lead to the following conclusions.

- The thiadiazole derivatives **12d, 12c, 6g, 18b, 6c** and **6f** showed high antitumor activity, and the thiadiazole derivatives **6e, 12a, 18a** and **6a** revealed moderate antitumor activity, while the thiadiazole derivatives **12b, 6b** and **6d** exhibited poor antitumor activity.
- The ester group (C₀₂Et) at position 2 of the thiadiazole ring is necessary to have higher antitumor activity than the acetyl and the N-phenylcarboxamide (CONHPh) groups.
- The presence of chlorine or bromine group (electron-withdrawing groups) at the position 2 or 4 in the aryl moiety of the thiadiazole ring as in the compounds **12d, 12c, 6g, 18b, 6c** and **6f** increased the cytotoxic activity. Also, halogen at positions 2 or 4 had more cytotoxic activity than halogen at position 3.
- While presence of electron-donating groups such as methyl or methoxy at the position 4 as in the compounds **12b, 6b** and **6d** decreased the cytotoxic activity.



6a-g, 12a-d, and 18a,b

The influence of substituents on the observed biological activity:

a: Type of X = electron withdrawing groups > electron donating groups

FIGURE 2. Structure–activity relationship of tested compounds against liver carcinoma cell line (HepG2)

III. MATERIALS AND METHODS

3.1 Chemistry

3.1.1 General

Melting points were measured on an Electrothermal IA 9000 series digital melting point apparatus. IR spectra were measured on Pye Unicam SP 3300 and Shimadzu FTIR 8101 PC infrared spectrophotometers in potassium bromide discs. NMR spectra were measured on a Varian Mercury VX-300 NMR spectrometer operating at 300 MHz (^1H -NMR) and run in deuterated dimethylsulfoxide (DMSO-*d*₆). Chemical shifts were related to that of the solvent. Mass spectra were recorded on a Shimadzu GCMS-QP1000 EX mass spectrometer at 70 eV. Elemental analyses were measured by using a German made Elementar vario LIII CHNS analyzer. Antitumor activity of the products was measured at the Regional Center for Mycology and Biotechnology at Al-Azhar University, Cairo, Egypt. 2-(4-Methyl-2-phenylthiazole-5-carbonyl)-N-phenylhydrazine carbothioamide (**3**) [35], 5-(4-methyl-2-phenylthiazol-5-yl)-1,3,4-oxadiazole-2(3*H*)-thione (**9**) [35], hydrazoneyl halides **4a-g**, **10a-d** and **16a,b** [36], and ethyl 5-hydrazono-4-phenyl-4,5-dihydro-1,3,4-thiadiazole-2-carboxylate (**15**) [37] were prepared as reported in the respective literature.

3.2 Synthetic Procedures

3.2.1 Synthesis of 1,3,4-thiadiazole derivatives (6a-g, 12a-d and 18a,b).

General procedure. A mixture of compound **3** (0.368 g, 1 mmol) and the appropriate hydrazoneyl chlorides **4a-g** or **10a-d** or **16a,b** (1 mmol) in ethanol (20 mL) containing triethylamine (0.1 g, 1 mmol) was refluxed for 4-6 h. (monitored by TLC). The formed solid product was filtered, washed with methanol, dried and recrystallized from the proper solvents to afford products **6a-g**, **10a-d** and **18a,b**, respectively. The physical constants and spectral data of the obtained products are listed below:

3.2.1.1 N'-(5-Acetyl-3-phenyl-1,3,4-thiadiazol-2(3*H*)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide (6a).

Yellow solid (73%); m.p. 163-165°C (EtOH); IR (KBr) ν 3317 (NH), 3038, 2951 (CH), 1701, 1647 (2C=O), 1593 (C=N) cm^{-1} ; ^1H -NMR (DMSO-*d*₆) δ 2.44 (s, 3H, CH₃), 2.74 (s, 3H, CH₃), 6.92-8.00 (m, 10H, ArH), 11.19 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 435 (M⁺, 10), 381 (13), 274 (56), 118 (31), 92 (100), 65 (38). Anal. Calcd. for C₂₁H₁₇N₅O₂S₂ (435.52): C, 57.91; H, 3.93; N, 16.08. Found C, 57.86; H, 3.84; N, 16.00%.

3.2.1.2 N'-(5-Acetyl-3-(p-tolyl)-1,3,4-thiadiazol-2(3*H*)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide (6b).

Yellow solid (75%); m.p. 149-151°C (EtOH); IR (KBr) ν 3334 (NH), 3019, 2920 (CH), 1699, 1648 (2C=O), 1597 (C=N) cm^{-1} ; ^1H -NMR (DMSO-*d*₆) δ 2.31 (s, 3H, CH₃), 2.44 (s, 3H, CH₃), 2.73 (s, 3H, CH₃), 6.98-7.89 (m, 9H, ArH), 11.18 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 449 (M⁺, 45), 372 (54), 200 (27), 104 (36), 80 (100), 64 (35). Anal. Calcd. for C₂₂H₁₉N₅O₂S₂ (449.55): C, 58.78; H, 4.26; N, 15.58. Found C, 58.65; H, 4.17; N, 15.46%.

3.2.1.3 N'-(5-Acetyl-3-(4-chlorophenyl)-1,3,4-thiadiazol-2(3*H*)-ylidene)-4-methyl-2-phenyl-thiazole-5-carbohydrazide (6c).

Brown solid (75%); m.p. 171-173°C (EtOH); IR (KBr) ν 3325 (NH), 3013, 2926 (CH), 1698, 1655 (2C=O), 1594 (C=N) cm^{-1} ; ^1H -NMR (DMSO-*d*₆) δ 2.45 (s, 3H, CH₃), 2.76 (s, 3H, CH₃), 6.93-7.96 (m, 9H, ArH), 11.25 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 471 (M⁺+2, 14), 469 (M⁺, 45), 396 (57), 200 (17), 80(100), 64 (89). Anal. Calcd. for C₂₁H₁₆ClN₅O₂S₂ (469.97): C, 53.67; H, 3.43; N, 14.90. Found C, 53.52; H, 3.37; N, 14.82%.

3.2.1.4 N'-(5-Acetyl-3-(4-methoxyphenyl)-1,3,4-thiadiazol-2(3*H*)-ylidene)-4-methyl-2-phenyl-thiazole-5-carbohydrazide(6d).

Brown solid (68%); m.p. 143-145°C (EtOH); IR (KBr) ν 3328 (NH), 3031, 2923 (CH), 1697, 1653 (2C=O), 1596 (C=N) cm^{-1} ; ^1H -NMR (DMSO-*d*₆) δ 2.45 (s, 3H, CH₃), 2.75 (s, 3H, CH₃), 3.76 (s, 3H, OCH₃), 6.99-7.99 (m, 9H, ArH), 11.29 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 465 (M⁺, 39), 334 (87), 200 (63), 122 (80), 77 (100), 64 (45). Anal. Calcd. for C₂₂H₁₉N₅O₃S₂ (465.55): C, 56.76; H, 4.11; N, 15.04. Found C, 56.63; H, 4.04; N, 14.95%.

3.2.1.5 N'-(5-Acetyl-3-(3-chlorophenyl)-1,3,4-thiadiazol-2(3H)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide(6e).

Yellow solid (70%); m.p. 166-168°C (EtOH); IR (KBr) ν 3431(NH), 3025, 2932 (CH), 1698, 1659 (2C=O), 1593 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.44 (s, 3H, CH_3), 2.66 (s, 3H, CH_3), 6.98-7.90 (m, 9H, ArH), 11.23 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 471 ($\text{M}^+ + 2$, 10), 469 (M^+ , 34), 334 (46), 200 (28), 132 (48), 80 (100), 64 (68). Anal. Calcd. for $\text{C}_{21}\text{H}_{16}\text{ClN}_5\text{O}_2\text{S}_2$ (469.97): C, 53.67; H, 3.43; N, 14.90. Found C, 53.60; H, 3.36; N, 14.79%.

3.2.1.6 N'-(5-Acetyl-3-(4-bromophenyl)-1,3,4-thiadiazol-2(3H)-ylidene)-4-methyl-2-phenyl thiazole-5-carbohydrazide (6f).

Brown solid (73%); m.p. 160-162°C (EtOH); IR (KBr) ν 3429 (NH), 3012, 2924 (CH), 1696, 1654 (2C=O), 1594 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.44 (s, 3H, CH_3), 2.65 (s, 3H, CH_3), 6.95-7.94 (m, 9H, ArH), 11.25 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 516 ($\text{M}^+ + 2$, 51), 514 (M^+ , 53), 325 (76), 172 (44), 91 (80), 80 (100), 64 (47). Anal. Calcd. for $\text{C}_{21}\text{H}_{16}\text{BrN}_5\text{O}_2\text{S}_2$ (514.42): C, 49.03; H, 3.14; N, 13.61. Found C, 48.93; H, 3.12; N, 13.53%.

3.2.1.7 N'-(5-Acetyl-3-(2,4-dichlorophenyl)-1,3,4-thiadiazol-2(3H)-ylidene)-4-methyl-2-phenyl thiazole-5-carbohydrazide(6g).

Brown solid (77%); m.p. 181-183°C (EtOH/dioxane); IR (KBr) ν 3318 (NH), 3088, 2926 (CH), 1699, 1671 (2C=O), 1597 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.47 (s, 3H, CH_3), 2.67 (s, 3H, CH_3), 6.97-8.07 (m, 8H, ArH), 11.19 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 504 (M^+ , 14), 407 (33), 161 (14), 80 (99), 64 (100). Anal. Calcd. for $\text{C}_{21}\text{H}_{15}\text{Cl}_2\text{N}_5\text{O}_2\text{S}_2$ (504.41): C, 50.00; H, 3.00; N, 13.88. Found C, 49.88; H, 2.92; N, 13.75%.

3.2.1.8 Ethyl 5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4-phenyl-4,5-dihydro-1,3,4- thiadiazole-2-carboxylate (12a).

Yellow solid (71%); m.p. 137-139 °C (EtOH); IR (KBr) ν 3432 (NH), 3035, 2923 (CH), 1749, 1659 (2C=O), 1597 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 1.20 (t, 3H, $J = 7.1$ Hz, CH_2CH_3), 2.74 (s, 3H, CH_3), 4.21 (q, 2H, $J = 7.1$ Hz, CH_2CH_3), 7.00-8.01 (m, 10H, ArH), 10.72 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 465 (M^+ , 27), 334 (50), 200 (34), 104 (40), 80 (100), 64 (37). Anal. Calcd. for $\text{C}_{22}\text{H}_{19}\text{N}_5\text{O}_3\text{S}_2$ (465.55): C, 56.76; H, 4.11; N, 15.04. Found C, 56.69; H, 4.03; N, 15.01%.

3.2.1.9 Ethyl 5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4-(p-tolyl)-4,5-dihydro-1,3,4- thiadiazole-2-carboxylate (12b).

Yellow solid (70%); m.p. 147-149°C (EtOH); IR (KBr) ν 3424 (NH), 3058, 2925 (CH), 1749, 1674 (2C=O), 1595 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 1.20 (t, 3H, $J = 7.1$ Hz, CH_2CH_3), 2.26 (s, 3H, CH_3), 2.76 (s, 3H, CH_3), 4.19 (q, 2H, $J = 7.1$ Hz, CH_2CH_3), 7.00-8.02 (m, 9H, ArH), 10.73 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 479 (M^+ , 20), 367 (25), 251 (18), 80 (85), 64 (100). Anal. Calcd. for $\text{C}_{23}\text{H}_{21}\text{N}_5\text{O}_3\text{S}_2$ (479.57): C, 57.60; H, 4.41; N, 14.60. Found C, 57.49; H, 4.33; N, 14.51%.

3.2.1.10 Ethyl 4-(4-chlorophenyl)-5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4,5- dihydro-1,3,4- thiadiazole-2-carboxylate(12c).

Yellow solid (73%); m.p. 167-169°C (EtOH/dioxane); IR (KBr) ν 3340 (NH), 3050, 2927 (CH), 1748, 1670 (2C=O), 1599 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 1.23 (t, 3H, $J = 7.1$ Hz, CH_2CH_3), 2.75 (s, 3H, CH_3), 4.22 (q, 2H, $J = 7.1$ Hz, CH_2CH_3), 7.02-7.96 (m, 9H, ArH), 10.77 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 501 ($\text{M}^+ + 2$, 13), 499 (M^+ , 45), 363 (39), 334 (100), 200 (35), 104 (30), 77 (50). Anal. Calcd. for $\text{C}_{22}\text{H}_{18}\text{ClN}_5\text{O}_3\text{S}_2$ (499.99): C, 52.85; H, 3.63; N, 14.01. Found C, 52.79; H, 3.60; N, 13.87%.

3.2.1.11 Ethyl 4-(2,4-dichlorophenyl)-5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4,5- dihydro-1,3,4- thiadiazole-2-carboxylate (12d).

Brown solid (75%); m.p. 173-175°C (EtOH/dioxane); IR (KBr) ν 3221 (NH), 3079, 2926 (CH), 1749, 1671 (2C=O), 1599 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 1.24 (t, 3H, $J = 7.1$ Hz, CH_2CH_3), 2.77 (s, 3H, CH_3), 4.23 (q, 2H, $J = 7.1$ Hz, CH_2CH_3), 7.08-8.13 (m, 8H, ArH), 10.77 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 534 (M^+ , 19), 449 (78), 223(100),

200 (54), 104 (58), 80 (85). Anal. Calcd. for $C_{22}H_{17}Cl_2N_5O_3S_2$ (534.44): C, 49.44; H, 3.21; N, 13.10. Found C, 49.29; H, 3.16; N, 13.02%.

3.2.1.12 5-(2-(4-Methyl-2-phenylthiazole-5-carbonyl)hydrazono)-N,4-diphenyl-4,5-dihydro-1,3,4- thiadiazole-2-carboxamide (18a).

Brown solid (76%); m.p. 176-178°C (EtOH/dioxane); IR (KBr) ν 3427, 3343 (2NH), 1672, 1653 (2C=O), 1597 (C=N) cm^{-1} ; 1H -NMR (DMSO- d_6) δ 2.75 (s, 3H, CH_3), 7.02-7.78 (m, 15H, ArH), 10.18(s, br, 1H, D_2O -exchangeable NH), 11.72 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 512 (M^+ , 8), 401 (00), 282 (10), 150 (22), 92 (26), 65 (29). Anal. Calcd. For $C_{26}H_{20}N_6O_2S_2$ (512.61): C, 60.92; H, 3.93; N, 16.39. Found C, 60.78; H, 3.85; N, 16.32%.

3.2.1.13 4-(2,4-Dichlorophenyl)-5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-N- phenyl-4,5-dihydro-1,3,4-thiadiazole-2-carboxamide (18b).

Brown solid (77%); m.p. 186-188°C (Dioxane); IR (KBr) ν 3429, 3337(2NH), 1692, 1656 (2C=O), 1591 (C=N) cm^{-1} ; 1H -NMR (DMSO- d_6) δ 2.76 (s, 3H, CH_3), 7.13-7.83 (m, 13H, ArH), 10.19(s, br, 1H, D_2O -exchangeable NH), 11.77 (s, br, 1H D_2O -exchangeable NH); MS m/z (%) 581 (M^+ , 38), 473 (64), 334 (72), 200 (35), 119 (65), 64 (100). Anal. Calcd. for $C_{26}H_{18}Cl_2N_6O_2S_2$ (581.50): C, 53.70; H, 3.12; N, 14.45. Found C, 53.62; H, 3.03; N, 14.32%.

3.2.2 Alternate synthesis of thiadiazole derivatives 6a and 18a

To a mixture of 5-(4-methyl-2-phenylthiazol-5-yl)-1,3,4-oxadiazole-2(3*H*)-thione (**9**) (0.275 g, 1 mmol) and hydrazonoyl chloride **4a** or **16a** (1 mmol) in absolute EtOH (25 mL), was added triethylamine (0.1g, 0.14 mL, 1 mmol). The reaction mixture was stirred at room temperature till methyl mercaptan ceased to evolve (3h). The solvent was evaporated and the residue was treated with ice/HCl mixture. The solid product was collected by filtration, washed with EtOH, dried, and recrystallized to give the respective compounds **6a**, and **18a**, that was identical in all respects (m.p., mixed m.p. and IR spectra) with that obtained from reaction of **4a** or **16a** with **3**.

3.2.3 Aternate synthesis of 12a

A mixture of ethyl 4-methyl-2-phenylthiazole-5-carboxylate (**1**)(0.247 g, 1 mmol) and ethyl 5-hydrazono-4-phenyl-4,5-dihydro-1,3,4-thiadiazole-2-carboxylate (**15**) (0.264g, 1 mmol) was refluxed in ethanol for 4 h. The solid product that separated was filtered off, washed with water and finally recrystallized to give the corresponding product, **12a** which was identical in all aspects (m.p., mixed m.p. and IR spectra) with those obtained from reaction of **3** with **10a**.

3.2.4 Synthesis of hydrazonoyl chlorides 19a-c

A mixture of 4-methyl-2-phenylthiazole-5-carbohydrazide (**2**) (2.33 g, 10 mmol) and the appropriate hydrazonoyl chlorides **4a-c** (10mmol) in ethanol (30 mL) was refluxed for 3 hr. The resulting solid product was collected and recrystallized from the proper solvent to give the corresponding products **19a-c**.

3.2.4.1 2-(2-(4-Methyl-2-phenylthiazole-5-carbonyl)hydrazono)-N'-phenylpropanehydrazonoyl chloride (19a).

Yellow solid (84%); m.p. 188-190°C (EtOH); IR (KBr) ν 3440, 3316(2NH), 3036, 2922(CH), 1640 (C=O), 1599 (C=N) cm^{-1} ; 1H -NMR (DMSO- d_6) δ 2.36 (s, 3H, CH_3), 2.76 (s, 3H, CH_3), 7.06-7.86 (m, 10H, ArH), 10.03 (s, br, 1H, D_2O -exchangeable NH), 10.57 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%): 413 (M^+ +2, 12), 411 (M^+ , 40), 375 (48), 202 (100), 174 (45), 71 (26). Anal. calcd for $C_{20}H_{18}ClN_5OS$ (411.91): C, 58.32; H, 4.40; N, 17.00. Found: C, 58.19; H, 4.37; N, 16.88%.

3.2.4.2 2-(2-(4-Methyl-2-phenylthiazole-5-carbonyl)hydrazono)-N'-(p-tolyl)propane- hydrazonoylchloride (19b).

Yellow solid (86%); m.p. 172-174 °C (EtOH);IR (KBr) ν 3437, 3313 (2NH), 3041, 2917 (CH), 1679(C=O), 1598 (C=N) cm^{-1} ; 1H -NMR (DMSO- d_6) δ 2.24 (s, 3H, CH_3), 2.34 (s, 3H, CH_3), 2.77 (s, 3H, CH_3), 7.08-7.99 (m, 9H, ArH), 10.06(s, br, 1H, D_2O -exchangeable NH), 10.59 (s, br, 1H, D_2O -exchangeable NH); MS m/z (%) 427 (M^+ +2, 10), 425 (M^+ , 33), 389 (26), 202 (81), 106 (100), 64 (66). Anal. calcd for $C_{21}H_{20}ClN_5OS$ (425.93): C, 59.22; H, 4.73; N, 16.44. Found: C, 59.18; H, 4.65; N, 16.37%.

3.2.4.3 N'-(4-Chlorophenyl)-2-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)propane- hydrazonoyl chloride (19c).

Yellow solid (87%); m.p. 194-196°C (DMF); IR (KBr) ν 3434, 3319 (2NH), 3044, 2926(CH), 1682 (C=O), 1593 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.37 (s, 3H, CH₃), 2.77 (s, 3H, CH₃), 7.08-7.99 (m, 9H, Ar-H), 10.06 (s, br, 1H, D₂O-exchangeable NH), 10.57(s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 446 (M⁺, 8), 283 (14), 202 (39), 104 (46), 80 (100), 64 (90). Anal. calcd for C₂₀H₁₇Cl₂N₅OS (446.35): C, 53.82; H, 3.84; N, 15.69. Found: C, 53.75; H, 3.79; N, 15.58%.

3.2.5 Synthesis of 1,3,4-thiadiazole derivatives 21a-c.

Method A: A mixture of compound **3** (0.368 g, 1 mmol) and the appropriate hydrazonoyl chlorides **19a-c** (1 mmol) in ethanol (20 mL) containing triethylamine (0.1 g, 1 mmol) was refluxed for 6 h.. The formed solid product was filtered,, washed with methanol, dried and recrystallized from the suitable solvents to give corresponding products **21a-c**.

3.2.5.1 4-Methyl-N'-(1-(5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4-phenyl-4,5-dihydro-1,3,4-thiadiazol-2-yl)ethylidene)-2-phenylthiazole-5-carbohydrazide (21a).

Yellow solid(74%); m.p. 162-164°C (EtOH); IR (KBr) ν 3421, 3307 (2NH), 3031, 2951 (CH), 1649 (C=O), 1596 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.34 (s, 3H, CH₃), 2.66 (s, 3H, CH₃), 2.76(s, 3H, CH₃), 6.97-8.14 (m, 15H, ArH), 10.18 (s, br, 1H, D₂O-exchangeable NH), 11.17 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 650 (M⁺, 34), 526 (30), 416 (60), 358 (28), 104 (55), 64 (100). Anal. Calcd for C₃₂H₂₆N₈O₂S₃ (650.80): C,59.06; H, 4.03; N, 17.22. Found C, 58.94; H, 4.01; N, 17.07%.

3.2.5.2 4-Methyl-N'-(1-(5-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-4-(p-tolyl)-4,5-dihydro-1,3,4-thiadiazol-2-yl)ethylidene)-2-phenylthiazole-5-carbohydrazide(21b).

Yellow solid (72%); m.p. 149-151°C (EtOH); IR (KBr) ν 3422, 3328 (2NH), 3053, 2929 (CH), 1647 (C=O), 1597 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.26 (s, 3H, CH₃),2.35 (s, 3H, CH₃), 2.65 (s, 3H, CH₃), 2.76(s, 3H, CH₃), 6.91-8.03 (m, 14H, ArH), 10.18 (s, br, 1H, D₂O-exchangeable NH), 11.14 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 664 (M⁺, 35), 553 (60), 334 (19), 202 (65), 104 (85), 64 (100). Anal. Calcd for C₃₃H₂₈N₈O₂S₃ (664.82): C, 59.62; H, 4.25; N, 16.85. Found C, 59.47; H, 4.17; N, 16.79%.

3.2.5.3 N'-(3-(4-Chlorophenyl)-5-(1-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-ethyl)- 1,3,4-thiadiazol-2(3H)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide (21c).

Yellow solid (76%); m.p. 191-193°C (Dioxane); IR (KBr) ν 3424, 3312 (2NH), 3047, 2932 (CH), 1649 (C=O), 1599 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.33 (s, 3H, CH₃), 2.66 (s, 3H, CH₃), 2.77(s, 3H, CH₃), 6.90-8.11 (m, 14H, ArH), 10.13 (s, br, 1H, D₂O-exchangeable NH), 11.19 (s, br, 1H, D₂O-exchangeable NH); MS m/z (%) 686 (M⁺+2, 8), 684 (M⁺, 26), 513 (53), 368 (39), 257 (17), 104 (25), 64 (100). Anal. Calcd for C₃₂H₂₅ClN₈O₂S₃ (685.24): C, 56.09; H, 3.68; N, 16.35. Found C, 56.02; H, 3.58; N, 16.22%.

Method B: A mixture of 4-methyl-2-phenylthiazole-5-carbohydrazide (**2**) (0.233 g, 1 mmol) and the appropriate 1,3,4-thiadiazoles **6a-c** (1 mmol) in ethanol (10 mL) was refluxed for 4h, allowed to cool and the solid product that formed was filtered off, washed with EtOH, dried and recrystallized from the proper solvent to give the corresponding product, **21a-c** which were identical in all aspects (m.p., mixed m.p. and IR spectra) with those obtained from reaction of **3** with **19a-c**.

3.2.6 General procedure for the synthesis of thiazole derivatives 23, 25, 27, and 29

A mixture of compound **3**(0.368 g, 1 mmol) and the appropriate α -halo-compounds namely, 3-chloropentane-2,4-dione (**22**), 2-chloro-3-oxo-N-phenylbutanamide (**24**), 2-bromo-1-phenyl ethanone (**26**) and ethyl 2-chloroacetate (**28**) (1 mmol for each) in ethanol (20 mL) containing triethylamine (0.1 g, 1 mmol) was refluxed for 4-6 h. (monitored by TLC The solid product was filtered, washed with water, dried and recrystallized from the proper solvent to give the corresponding thiazole derivatives **23**, **25**, **27**and **29**, respectively.

3.2.7 N'-(5-Acetyl-4-methyl-3-phenylthiazol-2(3H)-ylidene)-4-methyl-2-phenylthiazole-5- carbohydrazide (23).

Yellow solid (78%); m.p. 155-157°C (EtOH); IR (KBr) ν 3432 (NH), 3036, 2993 (CH), 1695, 1648 (2C=O), 1590 (C=N) cm^{-1} ; $^1\text{H-NMR}$ (DMSO- d_6) δ 2.32 (s, 3H, CH₃),2.46 (s, 3H, CH₃), 2.77 (s, 3H, CH₃), 6.91-7.86 (m, 10H, ArH), 10.61 (s, br,

¹H, D₂O-exchangeable NH); MS *m/z* (%) 448 (M⁺, 57), 246 (60), 176 (35), 104 (80), 77 (100). Anal. Calcd for C₂₃H₂₀N₄O₂S₂ (448.56): C, 61.59; H, 4.49; N, 12.49. Found C, 61.48; H, 4.36; N, 12.37%.

3.2.7.1 Methyl-2-(2-(4-methyl-2-phenylthiazole-5-carbonyl)hydrazono)-N-3-diphenyl-2,3-dihydrothiazole-5-carboxamide (25).

Yellow solid (79%); m.p. 182-84°C (DMF); IR (KBr): ν 3435, 3176(2NH), 3030, 2928(CH), 1671, 1649 (2C=O), 1594 (C=N) cm⁻¹; ¹H-NMR (DMSO-*d*₆) δ 2.36 (s, 3H, CH₃), 2.76(s, 3H, CH₃), 6.97-7.73 (m, 15H, ArH), 10.46 (s, br, 1H, D₂O-exchangeable NH), 11.72 (s, br, 1H, D₂O-exchangeable NH); MS *m/z* (%) 525 (M⁺, 7), 447 (16), 334 (100), 200 (59), 77 (89). Anal. Calcd for C₂₈H₂₃N₅O₂S₂ (525.64): C, 63.98; H, 4.41; N, 13.32. Found C, 63.84; H, 4.30; N, 13.28%.

3.2.7.2 N'-(3,4-Diphenylthiazol-2(3H)-ylidene)-4-methyl-2-phenylthiazole-5-carbohydrazide(27).

Yellow solid (70%); m.p. 174-178°C (EtOH); IR (KBr) ν 3369(NH), 3047, 2926(CH), 1648 (C=O), 1594 (C=N) cm⁻¹; ¹H-NMR (DMSO-*d*₆) δ 2.75 (s, 3H, CH₃), 7.03 (s, 1H, thiazole-H₅), 7.35-8.02 (m, 15H, ArH), 10.73 (s, br, 1H, D₂O-exchangeable NH); MS *m/z* (%) 468 (M⁺, 25), 334 (100), 200 (40), 104 (69), 64(65). Anal. Calcd for C₂₆H₂₀N₄OS₂ (468.59): C, 66.64; H, 4.30; N, 11.96. Found C, 66.55; H, 4.21; N, 11.79%.

3.2.7.3 4-Methyl-N'-(4-oxo-3-phenylthiazolidin-2-ylidene)-2-phenylthiazole-5-carbohydrazide (29).

Yellowish-white solid (72%); m.p. 192-194°C (Dioxane); IR (KBr) ν 3331(NH), 3036, 2926 (CH), 1726, 1648 (2C=O), 1596 (C=N) cm⁻¹; ¹H-NMR (DMSO-*d*₆) δ 2.65 (s, 3H, CH₃), 4.23 (s, 2H, thiazolone-CH₂), 7.40-7.88 (m, 10H, ArH), 10.82 (s, br, 1H, D₂O-exchangeable NH); MS *m/z* (%) 408 (M⁺, 65), 334 (18), 202 (100), 104 (86), 64 (69). Anal. Calcd for C₂₀H₁₆N₄O₂S₂ (408.50): C, 58.80; H, 3.95; N, 13.72. Found C, 58.68; H, 3.84; N, 13.64%.

3.3 Evaluation of the antitumor activity using Viability assay:

Human hepatocellular carcinoma (HepG2) cell line was obtained from the American Type Culture Collection (ATCC, Rockville, MD). The cells were grown on RPMI-1640 medium supplemented with 10% inactivated fetal calf serum and 50 μ g/mL gentamycin. The cells were maintained at 37 °C in a humidified atmosphere with 5% carbon dioxide and were subcultured 2 to 3 times a week. Potential cytotoxicity of the tested compounds was evaluated on tumor cells using the reported method of Gangadevi and Muthumary [43]. The cells were grown as monolayers in growth RPMI-1640. The monolayers of 10⁴ cells adhered at the bottom of the wells in a 96-well microtiter plate incubated for 24 h at 37 °C in a humidified incubator with 5% carbon dioxide. The monolayers were then washed with sterile phosphate buffered saline (0.01 M pH 7.2) and simultaneously the cells were treated with 100 μ L from different dilutions of tested sample in fresh maintenance medium and incubated at 37 °C. A control of untreated cells was made in the absence of tested sample. Positive controls containing doxorubicin drug was also tested as reference drug for comparison. Six wells were used for each concentration of the test sample. Every 24 h the observation under the inverted microscope was made. The number of the surviving cells was determined by staining the cells with crystal violet [44] followed by cell lysing using 33% glacial acetic acid and read the absorbance at 590 nm using microplate reader (SunRise, TECAN, Inc, USA) after well mixing. The absorbance values from untreated cells were considered as 100% proliferation. The number of viable cells was determined using microplate reader as previously mentioned before and the percentage of viability was calculated as $[1-(\text{ODt}/\text{ODc})] \times 100\%$ where ODt is the mean optical density of wells treated with the tested sample and ODc is the mean optical density of untreated cells. The relation between surviving cells and drug concentration is plotted to get the survival curve of each tumor cell line after treatment with the specified compound. The 50% inhibitory concentration (IC₅₀), the concentration required to cause toxic effects in 50% of intact cells, was estimated from graphic plots.

IV. CONCLUSIONS

In this context, a series of novel thiazoles and 1,3,4-thiadiazoles bearing thiazole were synthesized. The structure of the newly prepared compounds was established based on both elemental analysis and spectroscopic data and by an alternative method wherever possible. Moreover, the mechanisms of formation of the title compounds were discussed. Some of the synthesized compounds were evaluated for their anti-cancer activity against the human hepatocellular carcinoma (HepG2) cell line. The results showed that the thiadiazole derivatives **12d**, **12c**, **6g**, **18b**, **6c** and **6f** having IC₅₀ values 0.82, 0.91, 1.06, 1.25, 1.29 and 1.88 μ M, respectively, were found to be the highly active compounds of the prepared series. Based on the experimental results of the antitumor activity, the structure-activity relationships were discussed.

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AUTHOR CONTRIBUTIONS

S.M.G. designed research; S.M.G., M.R.A. and N.A.K. performed research and analyzed the data; S.M.G., N.A.K. and Y.N.M. wrote and approved the final manuscript.

CONFLICTS OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this paper.

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